

## Editorial: microbes vs. metals: harvest and recycle

Microbes and metals are intricately linked in a complex relationship. Many microbial pathways rely on metals for functionality, including enzymatic machinery (co-factors in key enzymes), dissimilatory reduction in energy generation (as alternative electron acceptors in anaerobic respiration) and biomineralization. Some metals share very close physical-chemical properties, thus sensing and incorporating the right metal in enzymes can be a finely-tuned, tightly regulated process (Waldron and Robinson 2009). In environments where essential metals have limited bioavailability microbes have developed high-affinity chelators (metallophores) or are capable of changing the redox conditions at the microscale level to solubilize them. The production of cellular energy using the redox-state transformations of metals/metalloids is an ancient strategy some bacteria and archaea use to sustain growth under nutrient-poor and sometimes extreme conditions (Stolz and Oremland 1999; Gescher and Kappler 2012; Staicu and Barton 2017; Wells et al. 2020). Biominerals are synthesized by microbes to alleviate metal stress and to serve diverse ecological functions (e.g. magnetotactic bacteria synthesize particles of single-domain magnetite *de novo*, providing them the ability to sense and orient in the ambient geomagnetic field). Metals can be toxic and microbes have developed sophisticated resistance mechanisms to counteract this type of stress. Strategies include ion specific efflux pumps, sequestration in poorly soluble minerals and redox change reactions (Nies 1999; Ni et al. 2015).

Metals are also key to human health, culture and industrialization. The typical multivitamin supplements contain a host of metals including selenium, copper, nickel, zinc, manganese and molybdenum, again a reflection of the presence of metalloenzymes and their co-factors. Metals have played a crucial role in human history (e.g. iron age and bronze age), and today more than ever, our daily life cannot be sustained without metals, from energy production and storage, transport, building and maintaining infrastructure or to technology. Mobile phones, which have revolutionized how people communicate, contain iron, nickel, gold, silver, platinum, palladium, aluminum, tin and copper (Sullivan 2006). While some metals are still abundant, the projected demand for others will outstrip supply in the foreseeable future, thus being listed as critical raw materials (CRMs) by various agencies and by the European Commission (<https://ec.europa.eu/>). In contrast to the renewable energy, metals are not at a practical time scale, therefore their use should be sustainable. Further, there is growing concern over the proliferation of 'e-waste' (i.e. discarded electronic devices, computers and cell phones). This means new strategies have to be implemented

for efficient recovery, reuse and for metal extraction from low-grade resources (e.g. metalliferous soils, metallurgical slags and mine waste; Vidal et al. 2017; Kissler et al. 2020). Microbes and their capacity to transport metals can be of great use in such a context as this approach offers a less energy-intensive and also a less environmentally-degrading alternative to processes such as hydro- and pyrometallurgy. Biomining and bioleaching, the recovery of metals using microbial metabolism, are successfully used in industrial operations for the extraction of copper, gold and uranium (Jerez 2017). This strategy can be applied to both metal-rich deposits as well as to low-grade metal wastes/slags resulted from past industrial activities and discarded in the environment (Potysz and Kierczak 2019). Another unexploited source is the metal-rich wastewater generated by numerous industrial activities. Such sources, in the framework of the growing circular economy paradigm shift, started to be regarded as a resource rather than a waste material. These industrial streams can be treated using a microbially-mediated bioremediation approach, coupled with metal recovery (Puyol et al. 2016). It may also be possible to recover metals from printed circuit boards of computers and cell phones (Argumedo-Delira, Diaz-Martinez and Gomez-Martinez 2020). Thus, we thought it timely to have a themed issue 'Microbes vs Metals: Harvest and Recycle'.

We have assembled a group of papers with a range of topics covering various chemical elements such as antimony, arsenic, copper, gold, iron, lead, manganese, selenium, tellurium, thallium and alloys (steel). They include contributions on metal resistance (Cyriaque et al. 2020; Sanyal, Reith and Shuster 2020; Yasir et al. 2020; Butz et al. 2021; Xu et al. 2020), biomineralization (Staicu et al. 2020), microbial ecology (Sjöberg et al. 2020; Taleski et al. 2020; Garrison and Field 2021; Zhang et al. 2021), as well as several review articles (Oremland 2020; Wells and Stolz 2020; Giachino et al. 2021; Mergeay and van Houdt 2021). Overall, the above-mentioned contributions demonstrate the complex relationship microbes have with chemically-diverse elements from the periodic table, from transition and precious metals to metalloids and alloys. Either as pure cultures or as mixed microbial communities, microbes are able to colonize and modify diverse environments, thus impacting the mobility and biogeochemical cycles of the elements in nature. From a technological perspective, microbes act as cleaning agents in industrially-polluted environments, being exploited in approaches that are less aggressive than the conventional treatment systems.

This themed issue is also dedicated to one of the pioneers of geomicrobiology, Professor Henry L. Ehrlich, whom we lost in 2020. Professor Ehrlich was known not only as an expert

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in microbial manganese metabolism, but also a great mentor, teacher and source of inspiration (Ghiorse 2020).

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